

Subject card

Subject name and code	Quantum chemistry, PG_00080828						
Field of study	Chemical Business						
Date of commencement of studies	October 2025		Academic year of realisation of subject			2025/2026	
Education level	Bachelor's studies		Subject group			Obligatory subject group in the field of study Subject group related to scientific research in the field of study	
Mode of study	full-time studies		Mode of delivery			at the university	
Year of study	1		Language of instruction			Polish	
Semester of study	2		ECTS credits			1.0	
Learning profile	academic		Assessment form			credit	
Conducting unit	Faculty of Chemistry -> Rector						
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. Iwona Anusiewicz				
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	0.0	15.0	0.0	0.0	0.0	15
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	15		2.0		8.0	25
Subject objectives	acquainting students with the basics of quantum mechanics and quantum chemistryacquainting students with the most important quantum chemistry methods allowing the prediction of their molecular structure, physicochemical properties, and reactivity.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[BCHINŽ_W02] Enumerates basic laws and theories in chemistry, physics and mathematics necessary to formulate and solve simple engineering tasks.	After the course, the students are capable of: formulating the basics of quantum mechanics, explaining simple physical problems solved by quantum mechanics, identifying the symmetry of the wave-function, formulating Pauli exclusion principle and Hund rules, explaining the most fundamental approximations utilized in quantum chemistry,	[SW4] test/exam - oral or written
	[BCHINŽ_U08] Uses the chemical nomenclature and engineering terminology properly.	After the course, the students are capable of: formulating the basics of quantum mechanics, explaining simple physical problems solved by quantum mechanics, identifying the symmetry of the wave-function, formulating Pauli exclusion principle and Hund rules, explaining the most fundamental approximations utilized in quantum chemistry, determining the multiplicity of a given molecular system	[SU4] test/exam - oral or written
	[BCHINŽ_W03] Describes the techniques of higher mathematics and IT tools necessary to describe and model chemical phenomena and technological processes.	After the course, the students are capable of: explaining the most important quantum chemistry methods.	[SW4] test/exam - oral or written
	[BCHINŽ_U02] Uses basic methods, techniques and tools in formulating and solving engineering tasks in the field of chemistry.	After the course, the students are capable of: choosing the most suitable computational method for solving the Schrödinger equation for a given molecular system.	[SU4] test/exam - oral or written
Subject contents	operators, eigenvectors and eigenvalues in Hilbert space, quantum numbers, orbitals, spin operators, symmetry of the wave-function, Slater determinants, evaluating electronic energy in Hartree-Fock method.		
Prerequisites and co-requisites	basic knowledge concerning physics, linear algebra, infinitesimal and integral calculus		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	Two written tests	51.0%	100.0%
Recommended reading	Basic literature	Molecular Quantum Mechanics (P. Atkins, R. Friedman), An Introduction to Theoretical Chemistry (J. Simons),Quantum Mechanics in Chemistry (J. Simons, J. Nicols).	
	Supplementary literature	Quantum Mechanics (A. Messiah), Ideas of Quantum Chemistry (L. Piela), Modern Quantum Chemistry (A. Szabo, N. Ostlund).	
	eResources addresses		
Example issues/ example questions/ tasks being completed	Is it possible to simultaneously measure the p(y) of the electron's momentum and the y of the position? Give an appropriate calculation.		
Work placement	Not applicable		

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