

		<b>ECTS code</b> 7.2.0624	
Name of unit administrating study Faculty of Chemistry			
Studies			
Туре	Form		
Bachelor	Full-time studies		
Forms of classes, the realization and number of hours			
A. Forms of classes, in accordance with the UG Rector's		classes - 45 h	
regulations		tutorial classes – 10 h	
lecture, laboratory classes			
B. The realization of activities			
in-class learning			
lasses)			
1055057	l		
Lang	Language of instruction		
Polish	Polish		
Form	and method of assessment and basic criteria for eva amination requirements	aluation or	
A. Fu	al evaluation, in accordance with the UG study reg	ulations	
lectur	lecture – exam		
labora	laboratory classes – course completion (with a grade)		
B. As	B. Assessment methods		
-	Written exam with open questions		
C. Th	basic criteria for evaluation or exam requirements		
• posi	• positive grade received in written exam composed of 5-10 open		
questi	questions covering issues listed in the course contents; answers to these questions will require solving tasks specified in educational outcomes:		
• to take passed;		exams	
		iust be	
Tutor	ıls:	_	
• pass	• passing two written colloquiums covering: (1) chemical structures and		
propert and pro-		structure	
memb	anes and nucleic acids:		
• each	negative grade should be improved at repeat colloquiv	ım.	
Labor	Laboratory classes:		
• posi	• positive grade received in 3 preliminary testes, that check knowledge		
requir	required to perform experiments during the classes; accomplishment of		
all pla	and experimental work (quality of laboratory work, all york and mode of work would be graded); analysis of a	bility to	
result	performed as written report.	Jotameu	
• to co	• to complete the laboratory course each negative grade must be		
impro	improved.		
	Study   Type   Bachelor   nber of hours   'ith the UG Rector's   'ith the UG Rector's   'lasses)   'lasses)   Langua Polish   Form a exa A. Fina lecture laborate   B. Asse   ''   C. The ' positi questio questio the grav ' to tak passed; <i>Tutoria</i> passin propert and pro membr ' each n Labora   irequire all plar team w results   ' to con improv	Studies   Form     Bachelor   Full-time studies     nber of hours   ECTS credits 3     ith the UG Rector's   classes - 45 h     utorial classes - 10 h   student's own work - 20 h     Total: 75 h - 3 ECTS   Total: 75 h - 3 ECTS     :lasses)   Form and method of assessment and basic criteria for evaluation requirements     A. Final evaluation, in accordance with the UG study regle   lecture - exam     laboratory classes - course completion (with a grade)   B. Assessment methods     -   -   Written exam with open questions     C. The basic criteria for evaluation or exam requirements   • positive grade received in written exam composed of 5-10 of questions covering issues listed in the course contents; answe questions will require solving tasks specified in educational of the grading scale would be adjusted to the range of all rated of the taboratory classes and <i>tutorials</i> m passed;     Tutorials:   • passing two written colloquiums covering: (1) chemical stra properties of amino acids, peptides and proteins (2) chemical and properties of monosaccharides, polysaccharides, lipids, or membranes and nucleic acids;     • passing two written colloquiums covering: (1) chemical stra properties of amino acids, peptides and proteins (2) chemical and properties of monosaccharides, polysaccharides, lipids, or membranes and nucleic acids;     • passing two written colloquiums covering: (2) chemical and	

Organic chemistry (bachelor level) Fundamentals of organic chemistry, skills to work in a chemical laboratory, knowledge of basic laboratory glassware, learning the principles of work in a biochemical laboratory



## Aims of education

- to acquaint students with all issues mentioned in the lecture contents;
- to introduce students to the basic endogenous organic compounds, their structure and functions;
- to acquaint students with basic metabolic pathways and relations between them;
- to teach students how to perform biochemical experiments using delivered instructions;
- to develop the ability to critically asses and interpret obtained experimental results and analysis of scientific sources;

A. Lecture: Energy-rich compounds, thermodynamics of biochemical reactions. Classification, structures and functions of enzymes. Mechanisms of enzyme catalysis. Carbohydrates, lipids and proteins – structures and functions. Biological membranes – structure and functions. Metabolic pathways: glycolysis, gluconeo-genesis, pyruvate decarboxylation, Krebs cycle, oxidative phosphorylation, glycogen metabolism, fatty ac-ids metabolism, amino acids metabolism, pentose phosphate pathway. Proteins G and signal transduction. Photosynthesis. DNA and RNA: replication, transcription, translation, PCR. Basics of genetic engineering. B. Tutorial: Chemical structure, physicochemical properties and biological functions of peptides, proteins, nucleic acids, phospholipids, mono- and polysaccharides.

C. The lab: completion of five experiments with the following topics: determination of activity of serine proteinases and their inhibitors using chromogenic substrates, determination of kinetic parameters of select-ed chromogenic substrate, separation of proteins by size-exclusion chromatography, phospholipid analysis by thin layer chromatography, determination of polysaccharides susceptibility to hydrolysis in low pH

## **Bibliography of literature**

- A. Literature required to pass the course J. M. Berg, J. L. Tymoczko, L. Stryer, "Biochemia", PWN, Warszawa 2009 Monographic works provided by assistants leading classes
- **B. Extracurricular readings** Various academic handbooks concerning biochemistry