

Course title Chemometria w analityce chemicznej / Chemometrics in chemical analytics		ECTS code 13.3.1010	
Name of unit administrating study Faculty of Chemistry			
Studies			
Field of study	Type	Form	
Chemistry	Bachelor	Full-time studies	
Teaching staff Prof. dr hab. Tomasz Puzyn			
Forms of classes, the realization and number of hours		ECTS credits 6	
A. Forms of classes, in accordance with the UG Rector's regulations lecture, auditorium classes, laboratory classes		classes - 60 h tutorial classes – 15 h student's own work – 75 h	
B. The realization of activities in-class learning		Total: 150 h - 6 ECTS	
C. Number of hours 60 h (30 h lecture, 15 h auditorium classes, 15 h laboratory classes)			
The academic cycle 2019/20 summer semester			
Type of course obligatory		Language of instruction Polish	
Teaching methods <ul style="list-style-type: none"> Lecture with multimedia presentation Students will conduct hands on exercises in the computational laboratory, based on the instructions prepared by the teacher Solving numerical tasks based on knowledge presented at the lecture and acquired by the student as part of their own studies 		Form and method of assessment and basic criteria for evaluation or examination requirements	
		A. Final evaluation, in accordance with the UG study regulations course completion (with a grade)	
		B. Assessment methods <ul style="list-style-type: none"> Written/oral final test (as described below) Colloquia that must be passed before each laboratory exercise Solving numerical tasks in computer laboratory Written reports summarizing results achieved during the lab exercises Determination of the final grade on the basis of partial assessments 	
		C. The basic criteria for evaluation or exam requirements Lecture: Written test consisting of several test questions and a few open questions (tasks) covering issues mentioned in the lecture content, auditorium and laboratory exercises. Final test: <ul style="list-style-type: none"> Written part (obligatory): single choice test with (about) 15 questions (1 point per question) plus (about) three open questions (5 points per each) – max. 30 points in total. Positive grade if the number of points > 50%. For students having between 40% and 50% from the written part, oral part is obligatory. Students with the number of points < 40% do not pass the exam. Oral part (obligatory for students having between 40% and 50% from the written part and facultative for students with > 50%): discussion on three problems related to the topic, selected by the teacher 	

	<ul style="list-style-type: none"> • Students are allowed passing the test twice (two attempts). • Students who obtained a written test in the first term of 51% and more, but want to increase their grade, may take oral part. The final mark in this case is the arithmetic average of the marks obtained on written and oral part. <p>Its obligatory to have a positive final grade from the lab exercises before passing the final test</p>
--	--

Required courses and introductory requirements

General chemistry

Mathematics (I sem., knowledge of the foundations of mathematics)

Aims of education

- Familiarizing the students with the possible application of chemometrics algorithms
- Acquiring by the students the skills of collecting, archiving and evaluating of the multivariable data
- Achieving basic skills in chemometric methods by the students (performing basic analyses and interpreting the results)
- Familiarizing the students with the available chemometric software.

Course contents

Lecture:

1. Introduction to chemometrics: specificity of multidimensional data; differences between statistics and chemometrics; area of interest in chemometrics; division of chemometric methods; review of the basic software (including MATLAB, Statistica, Origin, SPSS, QSARINS, KNIME).

2. Methods of analyzing the internal structure of the multidimensional chemical data: similarity of objects in the multivariable feature space; hierarchical cluster analysis (HCA) as an example of a similarity analysis method; principal component analysis (PCA) as an example of the projection search method. Examples of applying these methods in various areas of chemistry.

3. Modeling phenomena and processes with regression and classification methods: linear regression of single and multiple variables (LR and MLR), principal component regression (PCR) and partial least squares regression (PLS); linear discriminant analysis (LDA), a non-linear classifier of k-nearest neighbors (kNN); methods for selecting the optimal set of variables in the model (using the genetic algorithm); validation of regression and classification models. Examples of applying these methods in various areas of chemistry.

B. Auditorium classes:

1. Multidimensional data analysis - stages and goals
2. Hierarchical cluster analysis (HCA).
3. Principal Component Analysis (PCA).
4. Linear regression of single and multiple variables (LR / MLR).

C. Computer lab:

1. Introduction to chemometric calculations in the KNIEM software environment. The rules of working with a computer.
2. Hierarchical cluster analysis (HCA).
3. Principal Component Analysis (PCA).
4. Linear regression of single and multiple variables (LR / MLR).

Bibliography of literature

A. Literature required to pass the course:

- S. D. Brown, R. Tauler, B. Walczak (ed): Comprehensive chemometrics: Chemical and biochemical data analysis. Amsterdam: Elsevier, 2009
- R. Kramer: Chemometric techniques for quantitative analysis. New York: Marcel Dekker, Inc, 2005

B. Extracurricular readings:

- J. Leszczynski, A. Kaczmarek-Kedziera, T. Puzyn, M. G. Papadopoulos, H. Reis, M. Shukla (ed): Handbook of Computational Chemistry (2nd Edition). Springer 2016. Volume 5: Chemoinformatics, Puzyn T (ed.).
- T. Puzyn, J. Leszczynski, M. T. D. Cronin (ed): Recent Advances in QSAR Studies: Methods and Applications. Springer 2010. ISBN: 978-1-4020-9782-9.
- K. Roy, S. Kar, R. Narayan Das (ed): A Primer on QSAR/QSPR Modeling - Fundamental Concepts. Springer 2015. ISBN: 978-3-319-17281-1.

Knowledge

At the end of the course every student:

understands the need for reliable documentation of the results,
knows the basic division of chemometric methods, lists the use of particular groups of these methods in the analysis of chemical data;

knows basic software packages to be used for chemometric analyses;

knows the theoretical background (algorithm of operation) of the most important chemometric methods, including HCA, PCA, LR, MLR, PCR, PLS.

Skills

At the end of the course every student:

uses the KNIME environment for chemometric analyses;

correctly prepares data for further chemometric analysis;

performs HCA and PCA analyses and correctly interpret the obtained results;

develops regression model (LR / MLR method), validates the models correctly and applies the models for predictions.

Social competence

At the end of the course every student:

1. based on his knowledge and skills, critically evaluate the results of research using statistical methods (research, reports, etc.);
2. is convinced of the benefits of using a computer and introducing chemometric methods to his daily research practice;
3. understands the need of deeper learning of chemometric methods;
4. is aware that every numerical result is burdened with measurement uncertainty.