

| Course title | | | | ECTS code | | | | | | | |
|--|---------------------------|---|------------------|----------------------------------|--|------------------------------|--------------------|---------------------------|---|---|--|
| Wykład dyplomowy - Chemia i biochemia wybranych | | | | 13.3.0593 | | | | | | | |
| biomolekuł/Diploma lecture - C | hemistry and biochemistry | y of selected | | | | | | | | | |
| biomolecules | | | | | | | | | | | |
| Name of unit administrating study | | | | | | | | | | | |
| Faculty of chemistry | | | | | | | | | | | |
| Studies | | | | | | | | | | | |
| Field of study | Field of study Type | | | Form | | | | | | | |
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| Chemistry | Bachelor | Bachelor F | | 1-time studies | | | | | | | |
| Teaching staff Prof. dr hab. Krzysztof Rolka, prof. dr hab. Piotr Rekowski | | | | | | | | | | | |
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| Forms of classes, the realization and number of hours | | | EC15 creats | | | | | | | | |
| A. Forms of classes, in accordance with the UG Rector's | | | lecture 30 hours | | | | | | | | |
| regulations | | | | consultation 5 hours | | | | | | | |
| Lecture | | | | student's own work 15 hours | | | | | | | |
| B. The realization of activities | | | | | | | | | | | |
| lecture in the didactic room | | |] | TOTAL: 50 hours - 2 ECTS credits | | | | | | | |
| C. Number of hours | | | | | | | | | | | |
| 30 hours | | | | | | | | | | | |
| The academic cycle 2019/2020 summer semester | | | | | | | | | | | |
| | | | | | | | | | | | |
| | | Polish | 01 1115 | | | | | | | | |
| obligatory | | | | | | | | | | | |
| Teaching methods | | Form and method of assessment and basic criteria for evaluation or examination requirements | | | | | | | | | |
| Lecture with multimedia presentation | | A. Final evaluation, in accordance with the UG study regulations Course completion (with a grade) | | | | | | | | | |
| | | B. Assessment methods | | | | | | | | | |
| | | | | | | | | | | | |
| | | Written exam with open questions | | | | | | | | | |
| | | The basic criteria for evaluation | | | | | | | | | |
| | | Positive grade received in written exam composed of 6 open | | | | | | | | | |
| | | questions covering issues listed in the course contents; | | | | | | | | | |
| | | answers to these questions will require solving tasks | | | | | | | | | |
| | | specified in educational outcomes; the grade scale will be adjusted to the total number of points that could be obtained in the exam. Negative grade should be improved at repeat exam. The applied grading criteria will be in accordance | | | | | | | | | |
| | | | | | | | | with UG study regulations | | | |
| | | | | | | | | | • |) | |
| | | | | | | Required courses and introdu | ctory requirements | | | | |
| A. Formal requirements requirements The student should have completed a graduate study lectures | | | | | | | | | | | |

A. Formal requirements requirements The student should have completed a graduate study lectures ((bachelor level): "Organic chemistry", "Biochemistry" and "Polymer chemistry".

B. Prerequisites Basic knowledge in organic chemistry and biochemistry



Aims of education

- introduction students with all issues listed in the lecture program content,
- making students familiar with the basic groups of biomolecules their structures and functions,
- making students familiar with the basic methods of bioanalytical chemistry used for identification and
- quantitative and qualitative analysis of organic compounds occurring in living organisms.

Course contents

Analysis of biomolecules by liquid chromatography methods: thin layer chromatography, size exclusion, adsorption chromatography, separation in reverse-phase system, ion exchange chromatography, affinity chromatography. Gel and capillary electrophoresis. Mass spectrometry. Sequential analysis of nucleic acids and proteins. Hormones and neurotransmitters: structures and functions. Bacterial cell wall: structure and function. Antibiotics: classification and chemical structures. Ikosanoids: metabolism, chemical structures, biological functions. Xenobiotics. Fundamentals of chemical synthesis of peptides and nucleic acids. Chemical structures and biological functions of peptides, proteins, nucleic acids and polysaccharides.

Examples of protein (peptide) – nucleic acid interactions.

Bibliography of literature

- A. Literature required to pass the course
 - J. M. Berg, J. L. Tymoczko, L. Stryer, "Biochemia", PWN, Warszawa 2009.
- **B.** Extracurricular readings Monographic materials provided by the lecturers or chosen by students

Knowledge

- 1. Defines and describes chemical structures of selected macro- and biomolecules;
- 2. Describes the biological functions of naturally occurring compounds;
- 3. Describes the interactions between biomolecules;
- 4. Characterizes analytical techniques applied for analysis of endogenous organic compounds.

Skills

- 1. Uses chemical terminology necessary to present the content of the course;
- 2. Understands the role of naturally occurring compounds in processes taking place in living organisms;
- 3. Can search for information in specialist literature.

Social competence

- 1. Understands the need for continuous education;
- 2. Shows cautious criticism when acquiring knowledge, especially information coming from mass media;
- 3. Is aware of the necessity of fair and reliable work;
- 4. Can look at individual work with criticism.