

Course titleECTS codeWykład monograficzny - Zaawansowane metody13.3.0481elektrochemiczne/Monographic lecture - Advanced electrochemical13.3.0481

# Name of unit administrating study

Faculty of Chemistry

Studies			
Field of study	Туре	Form	
Chemistry	Master	Full-time studies	

#### **Teaching staff**

Dr Iwona Dabkowska

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Forms of classes, the realization and number of hours	ECTS credits		
A. Forms of classes, in accordance with the UG Rector's regulations  Lecture with multimedia presentation	Classes - 30 h Tutorial classes - 10 h Student's own work - 35 h TOTAL: 75 h - 3 ECTS		
B. The realization of activities			
In-class learning			
Number of hours			
lecture 30 h			

#### The academic cycle

2020/2021 summer semester

Type of course obligatory	Language of instruction Polish	
Teaching methods	Form and method of assessment and basic criteria for evaluation or examination requirements	
Lecture with multimedia presentation	A. Final evaluation, in accordance with the UG study regulations  Course completion (with grade)	
	B. Assessment methods test including open questions	
	C. The basic criteria for evaluation or exam requirements positive assessment: obtaining 51% of points from the written test consisting of 5 open questions (50%) and 20 test questions (50%) covering the issues listed in the lecture program content.	

### Required courses and introductory requirements

- a. Formal requirements completed courses in the field of instrumental analysis;
- b. Prerequisites the knowledge of basic methods of instrumental analysis,
  - the knowledge of electrochemical theory and laws.

#### Aims of education

- acquaintance with the history of development of electrochemical methods,
- becoming familiar with the theoretical foundations and measuring ranges of electroanalytical methods,
- introduction to selected measurement methods and imaging of various surfaces using electrochemical, optical and combined methods,
- presenting the application of electroanalytical methods to extend complex analytical problems,
- developing the skills of literature searches in the field of electrochemistry;



#### **Course contents**

(1)Theoretical concepts of electroanalytical methods: interfacial phenomena, the description of the diffusion of substances to the surface of the electrodes, the reversibility of electrode process, Fick's law, the Cotrell's law, modeling of electrodes' processes, electrodes' kinetics, mechanisms of electrodes' processes.

Electrochemical measurements in aqueous, non-aqueous and mixed environments.

(2) Voltammetric methods: chronoamperometry, cyclic voltammetry and normal pulse voltammetry. Electrochemical impedance spectroscopy. Voltammetric stripping.

- (3)Surface modification: Langmuir-Blodgett, self-organized film on the metal surface (SAM).
- (4)Techniques used to characterize monolayers: Raman spectroscopy, confocal and SERS.

Spectroelectrochemical measurements. The use of an atomic force microscope (AFM). Automation and computerization of analytical methods.

### Bibliography of literature

### A. Literature required to pass the course

#### **Primary literature:**

#### A.1.Literature used during classes:

- J. Bard, L. R. Faulkner Electrochemical methods, Wiley
- F. Scholz Electroanalytical methods, Guide to Experiments and Applications, Springer
- Z. Zoski Handbook of electrochemistry Elsevier
- Z. Galus Elektrochemiczne metody wyznaczania stałych fizykochemicznych, PWN, Warszawa
- Kisza Elektrochemia cz. I i II, WNT, Warszawa

### A.2. Literature for individual studies:

• W. Szczepaniak – Metody instrumentalne w analizie chemicznej, PWN, Warszawa

### Knowledge

#### **Students will:**

- 1. Define measurement principles and applications of various electrochemical methods.
- 2. Describe the theoretical principles of electrode processes and their mechanisms.
- 3. Recognize the types of surfaces obtained as a result of modifications.
- 4. Determine the physicochemical properties of the modified surface and how it interacts with other compounds.
- 5. Recognize the relationship between the type of measurement method and the accuracy of the data obtained.
- 6. Explain the principles of basic surface characterization techniques.
- 7. Describe the settings and operation of advanced electrochemical and spectroscopic devices.

## Social competence;

#### **Students will:**

- 1. Understand the need for further education.
- 2. Independently search for information in the world literature on the latest scientific reports in the field of advanced electroanalytical methods.