

2	KAPITAŁ LUDZKI Narodowa strategia spójności		nansowany ejską w rama ego Fundusz ecznego		
Course title				ECTS code	
Monographic lecture – Application of oxidation processe:			emistry	13.3.1170	
Name of unit admi					
Faculty of Chemis	strv				
Studies					
faculty	faculty field of study		drugiego sto	nnia	
Wydział Chemii	Chemia	form	stacjonarne		
		specialty		nedyczna, analityka i diagnostyka chemiczna, chemia i środowiska, chemia obliczeniowa	
		specialization			
Teaching staff					
-	Loop Chroursuréaldir prof. dr. ba	b Ewo Siedlaak	o, dr. bob la	anna Makawaka, profesar yazalni, dr.hah, Dariyaz	
-		D. Ewa Siedleck	a; or nad. Jo	anna Makowska, profesor uczelni; dr hab. Dariusz	
Wyrzykowski; dr Aleksandra Tesmar Forms of classes, the realization and number of hours				ECTS credits	
Forms of classes			3		
Lecture			s classes - 30 h		
Lecture The realization of activities				tutorial classes – 10 h	
			student's own work – 35 h		
classroom instruction					
Number of hours			Total: 75 h - 3 ECTS		
Lecture: 30 hours					
The academic cycl					
2023/2024 winter	semester				
Type of course		Langua	Language of instruction		
obligatory		polish			
Teaching methods		Form and method of assessment and basic criteria for eveluation or examination requirements			
- critical incident (Final ev	Final evaluation			
- discussion	Grade	Graded credit			
- group work - multimedia-base	Assess	Assessment methods			
- problem-focused	- assi	- assignment work – project or presentation			
			- graded course credit based on individual grades obtained during the		
		sem	ester		
		The bas	The basic criteria for evaluation		
		positive as	ssessment of	the presentation and activity in discussions covering the	subjec
	g required learning outcome				
Required courses	and introductory requireme	nts			
A. Formal requirem general chemistry, i	ents inorganic chemistry, analytical che	emistry, physical cł	nemistry, orga	nic chemistry	
B. Prerequisites					
Aims of education					
 familiarization with presentation of the developing the ab 	development of physicochemical the basic instrumental methods to e diversity of scientific works carrie ility to independently plan experim-	used in the charact ed out under the su nental work and sol	terization of te upervision of k lve problems	st substances in scientific works, KChOiN employees,	
• preparation for Ind	rependent selection of scientific lit	erature, leading co	insequently to	the preparation of a master's thesis	



Radicals and their types, radical reactions, the role of radical reactions in nature, classification of advanced oxidation processes (AOP), methods generation of radicals and other chemical species of the nature of oxidants and reducing agents, the use of AOP in water treatment, the use of AOP in water treatment, the use of reduction processes in the production of fuels, the use of radicals in medicine, the use of radicals in chemical synthesis, a review of the experimental methods used to study the antioxidant activity of natural compounds and synthetic compounds, including complex compounds, methods based on the HAT mechanism (hydrogen atom transfer), methods based on the mechanism SET (single electron transfer), electrochemical methods, factors determining the antioxidant activity of compounds

Bibliography of literature

Literature required to pass the course

Extracurricular readings

Literature provided by the teacher during the class

The learning outcomes (for the field of study and	Knowledge		
specialization)	Can classify radicals and radicals reactions; divides AOP methods depending on a method for generating hydroxyl radicals; lists the applications of radicals, redox chemical species and their reaction in environmental protection, medicine, chemical synthesis; can list methods commonly used in the analysis and diagnosis of radicals reactions; understands the description and the course of AOP processes; characterizes and understands the process of transfer electrons; explains the relationship between the structure of the relationship and its redox activity Skills		
	understands the interpretation of the results presented in scientific works; shows connections between the topics presented during the lecture and life; can indicate the application nature of the discussed and analyzed issues cases study; discusses the potential economic usefulness of the application of innovative methods that use radicals and chemical species of redox nature		
	Social competence		
	Discusses in a group collaborates with the colleagues, assumes various social roles (leader or performer etc.)		
Contact			
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