



**KAPITAŁ LUDZKI**  
NARODOWA STRATEGIA SPÓJNOŚCI

Projekt współfinansowany przez  
Unię Europejską w ramach  
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**UNIA EUROPEJSKA**  
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FUNDUSZ SPOŁECZNY



<b>Course title</b>		<b>ECTS code</b>	
Chemistry of environmental pollutants		13.3.1162	
<b>Name of unit administrating study</b>			
null			
<b>Studies</b>			
<b>faculty</b>	<b>field of study</b>	<b>type</b>	drugiego stopnia
Wydział Chemii	Chemia	<b>form</b>	stacjonarne
		<b>specjalty</b>	chemia i technologia środowiska
		<b>specialization</b>	wszystkie
<b>Teaching staff</b>			
dr hab. Łukasz Haliński; dr hab. Anna Białk-Bielińska, profesor uczelni; prof. dr hab. Piotr Stepnowski; dr hab. Magda Caban, profesor uczelni; prof. UG, dr hab. Monika Paszkiewicz; dr Joanna Dołżonek			
<b>Forms of classes, the realization and number of hours</b>		<b>ECTS credits</b>	
<b>Forms of classes</b>		4	
Laboratory classes, Lecture		classes 60 h	
<b>The realization of activities</b>		tutorial classes 20 h	
classroom instruction		student's own work 20 h	
<b>Number of hours</b>		TOTAL: 100 h - 4 ECTS	
Lecture: 30 hours, Laboratory classes: 30 hours			
<b>The academic cycle</b>			
2023/2024 winter semester			
<b>Type of course</b>		<b>Language of instruction</b>	
obligatory		polish	
<b>Teaching methods</b>		<b>Form and method of assessment and basic criteria for evaluation or examination requirements</b>	
<ul style="list-style-type: none"> <li>- conducting experiments</li> <li>- multimedia-based lecture</li> </ul>		<b>Final evaluation</b>	
		<ul style="list-style-type: none"> <li>- Graded credit</li> <li>- Examination</li> </ul>	
		<b>Assessment methods</b>	
		<ul style="list-style-type: none"> <li>- written exam with open questions</li> <li>- (mid-term / end-term) test</li> <li>- written exam (test)</li> <li>- graded course credit based on individual grades obtained during the semester</li> <li>- oral course credit</li> </ul>	
		<b>The basic criteria for evaluation</b>	

## C. The basic criteria for evaluation or exam requirements

## Lecture:

- pass the exam with open and closed questions concerning the whole course content

91-100%: 5.0

81-90%: 4.5

71-80%: 4.0

61-70%: 3.5

51-60%: 3.0

Less than 51% 2.0

## Laboratories:

- completed all tests

91-100%: 5.0

81-90%: 4.5

71-80%: 4.0

61-70%: 3.5

51-60%: 3.0

Less than 51% 2.0

The final grade will be calculated as a weighted average of the results of the final test covering all topics from the lab classes (40%), single tests from each class (40%) and laboratory reports (20%)

**Method of verifying required learning outcomes****Required courses and introductory requirements****A. Formal requirements**

Inorganic chemistry, organic chemistry, analytical chemistry, physical chemistry.

**B. Prerequisites**

Knowledge on the chemical structure and physicochemical properties of basic classes of organic and inorganic compounds. Basic skills in the chemical nomenclature, stoichiometry and its application, calculations of concentrations of compound solutions. Skills in using the laboratory glassware and basic analytical equipment. Knowledge on basic safety rules in the chemical laboratory.

**Aims of education**

- To provide students a clear understanding of the most important issues in the chemistry of environmental pollutants
- To familiarize students with the basic types of environmental pollutants, their transport, environmental fate and transformations
- To introduce students to principles of prediction of environmental fate of selected classes of pollutants
- To learn students how to independently estimate environmental risks associated with the use of certain substances basing on their chemical structure

**Course contents**

A. Topics of the lecture: Chemical pollution of the environment. Chemistry and physics of interaction of chemicals with components of environment. Environmental fate of selected pollutants: transport, stability and degradation. Global effects caused by the chemical pollution. Basic principles of environmental fate assessment of chemicals basing on structure-activity relationship.

B. Topics of lab classes: Determination of selected physicochemical parameters of pollutants using traditional and instrumental techniques.

Estimation of the impact of environmental conditions on environmental fate of chemicals. Determination of adsorption rate of chemicals to soils.

**Bibliography of literature**

Literature required to pass the course

A.1. literatur used during classes:

Alloway B.J., Ayres D.C. Chemiczne podstawy zanieczyszczenia środowiska, PWN, Warszawa, 1999.

Manahan S.E. Toksykologia środowiska. Aspekty chemiczne i biochemiczne, PWN, Warszawa, 2010.

Van Loon G.W., Duffy S.J. Chemia środowiska, PWN, Warszawa, 2008.

A.2. Literature for individual studies:

Manahan S.E. Toksykologia środowiska. Aspekty chemiczne i biochemiczne. PWN, Warszawa, 2010.

Van Loon G.W., Duffy S.J. Chemia środowiska. PWN, Warszawa, 2008.

Witkiewicz Z. Podstawy chromatografii. Wydawnictwa Naukowo-Techniczne, Warszawa, 2005.

Stepnowski P., Synak E., Szafranek B., Kaczyński Z. Monitoring i analityka zanieczyszczeń w środowisku. Wydawnictwo UG, 2010.

Extracurricular readings

Piotrowski J.K. (red.) Podstawy toksykologii. Kompendium dla studentów szkół wyższych. wyd. 2, WNT, Warszawa, 2008.

Pigon K. Chemia Fizyczna tom I. Wydawnictwo PWN, Warszawa, 2005.

Atkins P.W. Chemia fizyczna. PWN, Warszawa, 2001.	
<b>The learning outcomes (for the field of study and specialization)</b>	<b>Knowledge</b> Students define basic terms associated with risk assessment and chemical hazards. Students understand importance of structure-activity relationship of a substance in determining its environmental fate. Students are able to estimate the exposition of certain environment components on pollutants in association with the amounts of chemicals emitted and the way they are used. Students identify some characteristic types of the chemical structure that are responsible for certain physiochemical properties and biological activity. Students understand and characterize selected global effects, associated with the presence of chemical compounds in the environment.
	<b>Skills</b> <b>Social competence</b> Students understand the need of life-long learning in pollutant-related topics. Students show creativity in independent and team work. Students are able to estimate the impact of human activity on natural environment, in local and global scale, in a conscious way. Students are responsible for their own safety and safety of their co-workers; they are keeping safety measures when working with chemicals and analytical instruments.
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