



Projekt współfinansowany przez Unię Europejską w ramach Europejskiego Funduszu Społecznego



Course title E	ECTS code
Monographic lecture – Application of oxidation processes in chemistry	13.3.1170

#### Name of unit administrating study

Faculty of Chemistry

#### **Studies**

faculty	field of study	type	drugiego stopnia
Wydział Chemii	Chemia	form	stacjonarne
		specialty	chemia biomedyczna, chemia i technologia środowiska, analityka i
			diagnostyka chemiczna, chemia obliczeniowa
		specialization	wszystkie

# **Teaching staff**

prof. dr hab. inż. Lech Chmurzyński; dr Aleksandra Tesmar; dr hab. Joanna Makowska, profesor uczelni; dr hab. Dariusz Wyrzykowski; prof. dr hab. Ewa Siedlecka

Trytzykowoki, prof. di nas. Ewa elediceka		
Forms of classes, the realization and number of hours	ECTS credits	
Forms of classes	3	
Lecture	classes - 30 h	
The realization of activities	tutorial classes – 10 h	
classroom instruction	student's own work – 35 h	
Number of hours		
Lecture: 30 hours	Total: 75 h - 3 ECTS	

# The academic cycle

2023/2024 winter semester

Type of course	Language of instruction
obligatory	polish
Teaching methods  - critical incident (case) analysis  - discussion  - group work  - multimedia-based lecture  - problem-focused lecture	Form and method of assessment and basic criteria for eveluation or examination requirements
	Final evaluation
	Graded credit
	Assessment methods
	- assignment work – project or presentation
	- graded course credit based on individual grades obtained during the
	semester
	The basic criteria for evaluation
	positive assessment of the presentation and activity in discussions covering the subject

# Method of verifying required learning outcomes

# Required courses and introductory requirements

# A. Formal requirements

general chemistry, inorganic chemistry, analytical chemistry, physical chemistry, organic chemistry

#### **B. Prerequisites**

### Aims of education

presentation of the development of physicochemical research in solid and liquid phases over the last century,

- familiarization with the basic instrumental methods used in the characterization of test substances in scientific works,
- presentation of the diversity of scientific works carried out under the supervision of KChOiN employees,
- · developing the ability to independently plan experimental work and solve problems
- preparation for independent selection of scientific literature, leading consequently to the preparation of a master's thesis

#### **Course contents**

# Wykład monograficzny - Zastosowanie procesów utleniania w chemii #13.3.1170

Sylabusy - Centrum Informatyczne UG Dział Kształcenia



Radicals and their types, radical reactions, the role of radical reactions in nature, classification of advanced oxidation processes (AOP), methods generation of radicals and other chemical species of the nature of oxidants and reducing agents, the use of AOP in water treatment, the use of AOP in wastewater treatment, the use of reduction processes in the production of fuels, the use of radicals in medicine, the use of radicals in chemical synthesis, a review of the experimental methods used to study the antioxidant activity of natural compounds and synthetic compounds, including complex compounds, methods based on the HAT mechanism (hydrogen atom transfer), methods based on the mechanism SET (single electron transfer), electrochemical methods, factors determining the antioxidant activity of compounds

#### Bibliography of literature

Literature required to pass the course

Extracurricular readings

Literature provided by the teacher during the class

# The learning outcomes (for the field of study and specialization)

#### Knowledge

Can classify radicals and radicals reactions; divides AOP methods depending on a method for generating hydroxyl radicals; lists the applications of radicals, redox chemical species and their reaction in environmental protection, medicine, chemical synthesis; can list methods commonly used in the analysis and diagnosis of radicals reactions; understands the description and the course of AOP processes; characterizes and understands the process of transfer electrons; explains the relationship between the structure of the relationship and its redox activity

#### **Skills**

understands the interpretation of the results presented in scientific works; shows connections between the topics presented during the lecture and life; can indicate the application nature of the discussed and analyzed issues cases study; discusses the potential economic usefulness of the application of innovative methods that use radicals and chemical species of redox nature

#### Social competence

Discusses in a group collaborates with the colleagues, assumes various social roles (leader or performer etc.)

#### Contact

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