

Course titleECTS codeWykład monograficzny - Oddziaływania międzycząsteczkowe w
układach bionieorganicznych / Monographic lecture - Intermolecular13.3.0917

interactions in bioinorganic systems Name of unit administrating study

Faculty of Chemistry

Studies				
Field of study	Туре	Form		
Chemistry	Masters	Full-time studies		

Teaching staff

Prof. dr hab. Mariusz Makowski

FIOL UI HAU. IVIAHUSZ IVIAKOWSKI			
Forms of classes, the realization and number of hours	ECTS credits 3		
A. Forms of classes, in accordance with the UG Rector's	classes - 30 h		
regulations	tutorial classes – 10 h		
lecture	student's own work – 35 h		
B. The realization of activities			
in-class learning	Total: 75 h - 5 ECTS		
C. Number of hours	1000.701 02010		
30 h lecture			

The academic cycle

Second year, summer semester

Type of course obligatory	Language of instruction Polish
Teaching methods	Form and method of assessment and basic criteria for evaluation or examination requirements
Lecture with multimedia presentation	A. Final evaluation, in accordance with the UG study regulations course completion (with a grade)
	B. Assessment methods written test with open questions
	C. The basic criteria for evaluation or exam requirements Completion of the lecture based on obtaining a positive grade from a written test consisting of open questions covering the issues listed in the program contents. Passing criteria in accordance with the UG Studies Regulations.

Required courses and introductory requirements

None

Aims of education

Leading, through lectures, to understanding and grounding the basic concepts and concepts of intermolecular interactions in bioinorganic chemistry and to indicate the role they play in the chemical bases of selected biochemical processes.

Course contents

Theories of chemical bonds. Types of intermolecular interactions. The molecule and its surroundings. Intermolecular and intramolecular hydrogen bonds. Low energy intermolecular interactions in complex compounds. Factors influencing the strength of intermolecular interactions. Intermolecular interactions and physical properties. Theoretical and experimental evidence of the existence of hydrogen bonds. Elements of pharmacokinetics.

Bibliography of literature

A. Literature required to pass the course

- 1. P.A. Cox, Krótkie wykłady, chemia nieorganiczna, PWN, Warszawa, 2003.
- 2. F.A. Cotton, G. Wilkinson, P.L. Gaus, Chemia nieorganiczna, podstawy, PWN, Warszawa, 1995.

B. Extracurricular readings

- 1. N.N. Greenwood, A. Earnshaw, Chemistry of the elements, Pergamon, wyd. II, 2005.
- 2. C.E. Housecroft, A.G. Sharpe, Inorganic chemistry, Pearson, Prentice Hall, Ed I (2001), Ed II (2005) lub Ed III (2008);



- 3. S.J. Lippard, J.M. Berg, Podstawy chemii bionieorganicznej, PWN, Warszawa, 1998.
- 4. I.G. Kaplan, Intermolecular Interactions, chap. 1,2,5, Wiley, 2006.
- 5. P. Schuster, G. Zundel and C. Sandorfy, Eds., The Hydrogen Bond, Recent Developments in Theory and Experiments, North Holland., 1976.
- 7. Czasopisma wskazane przez prowadzącego zajęcia.

Knowledge

defines basic concepts in the chemistry of bioinorganic interactions;

lists and characterizes basic biological ligands

is able to describe the structure and functions of the most important bioinorganic systems;

Skills

can describe the structure and functions of the most important inorganic organic systems;

Social competence

Is interested in basic chemical processes taking place in the environment and follows literature reports indicated by the subject