

Course title			ECTS code	
Wykład monograficzny - Zaawan	nsowane metody	13.3.0481		
elektrochemiczne/Monographic lecture - Advanced electrochemical				
methods				
Name of unit administrating study				
Faculty of Chemistry				
Studies				
Field of study	Туре		Form	
Chemistry	Master F		ull-time studies	
Teaching staff				
Dr Iwona Dąbkowska				
Forms of classes, the realization and number of hours			ECTS credits	
			Classes - 30 h	
A. Forms of classes, in accordance with the UG Rector's			Tutorial classes - 10 h	
regulations			Student's own work - 35 h	
Lecture with multimedia presentation			TOTAL: 75 h - 3 ECTS	
B. The realization of activi	ties			
In close looming				
In-class learning				
Number of hours				
lecture 30 h				
The academic cycle Second year summer semester				
Type of course				
obligatory Polish		olish		
Teaching methods	Fe	orm and met	thod of assessment and basic criteria for evaluation or	
e e		examinati	examination requirements	
Lecture with multimedia presenta	ation	A Final evaluation in accordance with the UC study regulations		
		Course completion (with grade)		
B. Assessment methods				
test including open questions				
C. The basic criteria for evaluation or exam requirements				
	positive assessment: obtaining 51% of points from the			
written test consisting of 5 open questions (50%) ar				open questions (50%) and 20
test que lecture			estions (50%) covering the issues listed in the	
			program content.	
Required courses and introductory requirements				
a. Formal req	uirements completed c	ourses in th	he field of instrume	ental analysis:
b Prereguisites - the knowledge of basic methods of instrumental analysis				
the knowledge of electrochemical theory and laws				
- the knowledge of electrochemical theory and laws.				

Aims of education

- acquaintance with the history of development of electrochemical methods,
- becoming familiar with the theoretical foundations and measuring ranges of electroanalytical methods,
- introduction to selected measurement methods and imaging of various surfaces using electrochemical, optical and combined methods,
- presenting the application of electroanalytical methods to extend complex analytical problems,
- developing the skills of literature searches in the field of electrochemistry;



Course contents

(1)Theoretical concepts of electroanalytical methods: interfacial phenomena, the description of the diffusion of substances to the surface of the electrodes, the reversibility of electrode process, Fick's law, the Cotrell's law, modeling of electrodes' processes, electrodes' kinetics, mechanisms of electrodes' processes.

Electrochemical measurements in aqueous, non-aqueous and mixed environments.

(2)Voltammetric methods: chronoamperometry, cyclic voltammetry and normal pulse voltammetry. Electrochemical impedance spectroscopy. Voltammetric stripping.

(3)Surface modification: Langmuir-Blodgett, self-organized film on the metal surface (SAM).

(4)Techniques used to characterize monolayers: Raman spectroscopy, confocal and SERS.

Spectroelectrochemical measurements. The use of an atomic force microscope (AFM). Automation and computerization of analytical methods.

Bibliography of literature

A. Literature required to pass the course

Primary literature:

A.1.Literature used during classes:

- J. Bard, L. R. Faulkner Electrochemical methods, Wiley
- F. Scholz Electroanalytical methods, Guide to Experiments and Applications, Springer
- Z. Zoski Handbook of electrochemistry Elsevier
- Z. Galus Elektrochemiczne metody wyznaczania stałych fizykochemicznych, PWN, Warszawa
- Kisza Elektrochemia cz. I i II, WNT, Warszawa

A.2. Literature for individual studies:

• W. Szczepaniak - Metody instrumentalne w analizie chemicznej, PWN, Warszawa

Knowledge

Students will:

- 1. Define measurement principles and applications of various electrochemical methods.
- 2. Describe the theoretical principles of electrode processes and their mechanisms.
- 3. Recognize the types of surfaces obtained as a result of modifications.
- 4. Determine the physicochemical properties of the modified surface and how it interacts with other compounds.

5. Recognize the relationship between the type of measurement method and the accuracy of the data obtained.

6. Explain the principles of basic surface characterization techniques.

7. Describe the settings and operation of advanced electrochemical and spectroscopic devices.

Social competence;

Students will:

1. Understand the need for further education.

2. Independently search for information in the world literature on the latest scientific reports in the field of advanced electroanalytical methods.