

ECTS code Course title Chemia bionieorganiczna/Bioinorganic chemistry 13.3.1161 Name of unit administrating study Faculty of Chemistry **Studies** Field of study **Type Form** Chemistry Master Full-time studies Teaching staff Prof. dr hab. Mariusz Makowski Forms of classes, the realization and number of hours **ECTS** credits classes 45 h Tutorial classes 2 h Forms of classes, in accordance with the UG Rector's Student's own work 3 h regulations TOTAL: 50 h - 2 ECTS lecture, laboratory classes B. The realization of activities

In-class learning

C. Number of hours Lecture 15 h, laboratory classes 30 h

The academic cycle

Second year, winter semester

Type of course obligatory			Language of instruction Polish
Teaching methods Lectures presentations	including	multimodal	Form and method of assessment and basic criteria for evaluation or examination requirements
			A. Final evaluation, in accordance with the UG study regulations Course completion (with a grade)
			B. Assessment methods test
			C. The basic criteria for evaluation
			positive grade from written test consisting of 12-20 open questions comprising issues listed in the program content (lecture)

Required courses and introductory requirements

A. Formal requirements

none

B. Prerequisites

basic knowledge of inorganic and coordination chemistry

Aims of education

- familiarity with the problems occurring on the border of chemical, biological and medical sciences
- introduction of both basic and specialized knowledge of biochemistry (in particular, information about the role that bio-elements such as iron, copper, zinc, cobalt, manganese, nickel and chromium, play in living organisms)

Course contents

Bioinorganic chemistry - explanation of the term, foundations; Review of the most important groups of compounds (sugars, lipids, proteins and amino acids, vitamins - coenzymes, DNA / RNA) necessary for life; Biological demand for metals and inorganic compounds; The functions of metal ions in proteolysis. Methods of studies on bioinorganic compounds. Redox reactions with electron transfer in biological systems. Oxygen transfer and transport processes in cells. Circulation of nitrogen at the molecular level. Metal physiology. Medical chemistry of inorganic compounds. Environmental chemistry of bioinorganic compounds



Bibliography of literature

- A. Literature required to pass the course
 - A.2. Literature for individual studies:
 - L. Stephen, B. Jeremy Podstawy chemii bionieorganicznej
 - R. M. Roat-Malone Bioinorganic Chemistry: A Short Course
 - E. Ochiai Bioinorganic Chemistry: a survey
- B. Extracurricular readings

Bioinorganic Chemistry and Applications – science magazine

Knowledge

Student knows and understands the law, concepts and phenomena on the border of three areas: chemistry, biology and medicine.

Social competence

Student understands the need for further education. can formulate questions precisely to deepen understanding of a given topic or to find missing elements of reasoning; understands and appreciates the importance of intellectual honesty in own and other people's actions; act ethically; understands the need for popular presentation of selected issues in chemistry to non-specialists; can independently search for information in literature, including foreign language.